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Ch 9 Ionic And Covalent Bonding Ionic BondsCh 9 Ionic And Covalent Bonding Ionic Bonds - An Ionic Bond Is Formed By Electrostatic Attraction Between Cations (positive) And Anions (negative). - Some Elements, Primarily Metals Like Na (with $[\text{Ne}]3s^1$), Lose e^- 's Easily To 5th, 2024Ionic Bonds Properties Of Ionic Compounds (salts)Naming Binary Ionic Compounds Contain 2 Different Elements Name The Metal First, Then The Nonmetal With The Ending -ide. Use Name Of A Metal With A Fixed Charge These Are: Groups 1A, 2A, 3A And Ag $+$, Zn $^{2+}$, And Cd $^{2+}$ Examples: NaCl Sodium Chloride ZnI₂ Zinc Iodide Al₂O₃ Aluminum Oxide Lear 5th, 20247.2 Ionic Bonds And Ionic Compounds - Henry County SchoolsThe Ionic Compound Magnesium Chloride Contains Magnesium Cat-ions (Mg^{2+}) And Chloride Anions (Cl^-). In Magnesium Chloride, The Ratio Of Magnesium Cations To Chloride Anions Is 1:2 (one Mg^{2+} To Two Cl^-). So Its Formula Unit Is MgCl_2 . Because There Are Twice As Many Chloride Anions (each With A 1 2th, 2024.

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