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Bomb Calorimetry And Heat Of Combustion Where CH_0 Is The Heat Of Combustion Of Benzoic Acid (given As -6318 Cal G), M Is The Mass Of The Benzoic Acid Sample (0.9799 G), E_3 Is The Heat Of Combustion Of The Wire (calculable From The Heat Of Combustion Of The Wire, -2.3 Cal Cm And The Length Of The Wire, 15.1 Cm), And T Is The Previously Calculated Temperature Rise, 2.77252 C . Combining May 3th, 2024

Experiment 9: CALORIMETRY $\Delta T = \text{Change In Temperature Of The Solution} = T_{\text{Final}} - T_{\text{Initial}}$ In This Experiment Mass Of The Solution Will Be Calculated Using The Volume Of The Solution And Its Density (assumed To Be 0.997 G ML^{-1}) Heat Of Neutralization Is The Same Value As Q_{Soln} But With The Opposite Sign. The Solution Is Absorbing Heat So Q_{Soln} Is Positive. Apr 7th, 2024

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