

# Ch 10 Chemical Quantities Free Pdf Books

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COMPARING QUANTITIES Comparing Quantities

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Ch. 7: Chemical Reactions And Chemical Quantities

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 Reaction always Involves A Chemical Change Because  
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 Quantities 1-carbon 2-oxygen Thus One Mole Of CO<sub>2</sub>  
 Contains Three Times Avogadro's # Of Atoms.  
 Example: How Many Atoms Are In 2.12 Mole Of  
 Propane (C<sub>3</sub>H<sub>8</sub>)? Know: # Of Moles = 2.12 Mole C<sub>3</sub>H<sub>8</sub>  
 H<sub>8</sub> 1 Mole C<sub>3</sub>H<sub>8</sub> = 6.02 × 10<sup>23</sup> Molecules C<sub>3</sub>H<sub>8</sub> 1  
 Molecule C<sub>3</sub>H<sub>8</sub> = 11 Atoms ( 3 Carbon Atoms & 8

Hydrogen Atoms) Feb 7th, 2024  
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NOTES: 10.1-10.2 Chemical Quantities (The Mole / Molar ... Chemical Quantities - consider Another Word That Means A #: Measuring Doughnuts: A DOZEN! 1 Dozen = 12 Doughnuts (count) 1 Dozen = 500 G Doughnuts (mass) 1 Dozen = 1 Box Doughnuts (volume) Measuring Hydrogen (H<sub>2</sub> Gas): A MOLE! 1 Mole =  $6.02 \times 10^{23}$  H<sub>2</sub> Molecules (count) 1 Mo Jan 18th, 2024 Chapter 10 Chemical Quantities Workbook Answers Books File Access Free Chapter 10 Chemical

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