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Whole Numbers: $2 \text{NH}_3 + \frac{7}{2} \text{O}_2 \rightarrow 2 \text{NO}_2 + 3 \text{H}_2\text{O}$.
The Question Asks About The Amount Of Oxygen
Reacting With ONE Mole Of Ammonia, Thus Cut The $\frac{7}{2}$
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