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1-carbon 2-oxygen Thus One Mole Of CO 2 Contains Three Times Avogadro's # Of Atoms. Example: How Many Atoms Are In 2.12 Mole Of Propane (C 3 H 8)? Know: # Of Moles = 2.12 Mole C 3 H 8 1 Mole C 3 H 8 = 6.02 X 1023 Molecules C 3 H 8 1 Molecule C 3 H 8 = 11 Atoms (3 Carbon Atoms & 8 Hydrogen Atoms) 14th, 2024

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