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Important Values, Constants And Standards 3 2
Ionisation Energies (1st, 2nd, 3rd And 4th) Of Selected Elements In KJ Mol -1 4 ... 11 The Periodic Table Of Elements 17 . 3 Cambridge Pre-U Rev 5th, 2024

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+ H₂O + V³⁺ + H₂ + 2 OH⁻ What Would Alter The Composition Of The Equilibrium Mixture In Favour Of The V²⁺ Ions? A Adding An Acid B Adding A Reagent That Selectively Precipitates 2th, 2024

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(c) (i) Write A Chemical Equation Representing The Lattice Energy Of AgBr. (ii) Use The Following Data To Calculate A Value For The Lattice Energy Of AgBr(s).
First Ionisation Energy Of Silver = +731kJmol⁻¹ Electron Affinity Of Bromine = 325kJmol⁻¹ Enthalpy Change Of Atomisation Of Silver = +285kJmol⁻¹ Enthalpy Change Of Atomisation Of Bromine = +112kJmol⁻¹ ... 1th, 2024

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