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Calculating PH And POH Worksheet4) -A Solution Is Created By Measuring  $3.60 \times 10^3$  Moles Of NaOH And  $5.95 \times 10^{-4}$  Moles Of HCl Into A Container And Then Water Is Added Until The Final Volume Is 1.00 L. What Is The PH Of This Solution? Since There Is Both Acid And Base We Will Assume A 1 Mole Acid:1 Mole Base Ratio Of Neutralization. Apr 2th, 2024 PH And POH - Ms. Mogck's Classroom - Ms. Mogck's

ClassroomKEY Chemistry: PH And POH Calculations Part 1: Fill In The Missing Information In The Table Below. PH [ H 3 O 1+] POH [ OH1-] ACID Or BASE? 3.78  
 1.66 X 10<sup>-4</sup> M 10.22 6.03 X 10<sup>-11</sup> M Acid 3.41 3.89 X 10<sup>-4</sup> M 10.59 2.57 X 10<sup>-11</sup> M  
 Acid 8.81 1.55 X 10<sup>-9</sup> M 5.19 6.46 X 10<sup>-6</sup> M Base 8.69 2.04 X 10<sup>-9</sup> M 5.31 4.88 X  
 10<sup>-6</sup> M Base 8.46 3.47 X 10<sup>-9</sup> M 5.54 2.88 X 10<sup>-6</sup> M Base Mar 1th, 2024Basic  
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 A Shiny Metal, And Chlorine Is An Irritating Gas. But The Compound Formed From  
 These Two Elements, Sodium Chloride, Shows Different Characteristics As It Is A  
 Crystalline Solid, Vital For Biological Functions. Physical Classifi Cation Solid  
 Mixtures Ice Water Water Vapour Homogeneous May 3th, 2024.  
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 Decimal Is Absent Or Present. Atlantic Decimal Is Absent.Begin Counting On The  
 Atlantic Side (the Right). From RIGHT Of The Number, Find The First Non-zero Digit  
 And ... Feb 5th, 2024Chemistry Unit 2: Measurement And Calculations Name ...LT 2:  
 I Can Accurately Use Scientific Notation; I Can Use And Convert Correct SI Units Of  
 Measurement. Scientific Notation Notes Scientific Notation Can Be Expressed As A  
 Number Times Ten To A Power. PN X 10 N = Any Number From 1.00-9.99 P = Any

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