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General Information You Will Need A Periodic Table, Your Stoichiometry Notes,

And Appendix 12 For This Assignment. Sorry About The Lack Of Format. I'm In A Time Crunch. 123.88 G/mol 70.90 G/mol 137.32 ... 1th, 2024.

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KEYChemistry 11 Stoichiometry Review Worksheet  
Name:\_\_\_KEY\_\_\_ 33.  $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$

Actual yield = 1.45g Theoretical yield = 5.95g  
 $\text{KClO}_4 \times 1 \text{ mol} = 122.55 \text{ g}$   
 $\times 2 \text{ O}_2 \times 32.00 \text{ g} = 3.11 \text{ g O}_2$   
 $\% \text{ Yield} = \frac{1.45 \text{ g}}{3.11 \text{ g}} \times 100\% = 46.6\%$   
34.  $4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$

Percent purity = 62.8%? pure  $\text{FeCO}_3$  2th, 2024

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