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A) [Co(H<sub>2</sub>O)<sub>6</sub>]<sup>2+</sup> Co +2 E) [FeO<sub>4</sub>]<sup>2-</sup> Fe +6 I) K<sub>2</sub>[CoCl<sub>4</sub>] Co +2 B) [CrCl<sub>6</sub>]<sup>3-</sup> Cr +3  
F) [Mn(CN)<sub>6</sub>]<sup>4-</sup> Mn +2 J) K<sub>3</sub>[AuF<sub>6</sub>] Au +3 C) [Co(NH<sub>3</sub>)<sub>6</sub>]Cl<sub>2</sub> Co +2 G) [Ni(CO)<sub>4</sub>] Ni 0  
K) (NH<sub>4</sub>)<sub>2</sub>[IrCl<sub>6</sub>] Ir +4 D) [Co(NH<sub>3</sub>)<sub>5</sub>]Cl<sub>2</sub> Co +3 H) [Ni(EDTA)]<sup>2-</sup> Ni +2  
L) Na[Mn(CO)<sub>5</sub>] Mn -1 TASK 2 - Writing Half Equations A) 2I<sup>-</sup> → I<sub>2</sub> + 2e<sup>-</sup> B ...  
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Ethanol Has The Formula C<sub>2</sub>H<sub>5</sub>OH And Is Used As A Fuel (e.g. For Cars In Brazil).  
It Burns ... Jan 7th, 2024

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Metal Carbonate → Metal Oxide + Carbon Dioxide (on Heating) CaCO<sub>3</sub> → CaO + CO<sub>2</sub>

2 TASK 4 - WRITING BALANCED EQUATIONS 1) Balance The Following Equations.  
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