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2. Q. List Three Subatomic Particles Of An Atom. Compare Them On The Basis Of Relative Mass And Charge In A Tabular Form. Ans: Particle Symbol Charge Mass Electron  $e^{-}$   $9.1 \times 10^{-31}$  Kg Proton  $+p$  (H<sup>+</sup>)  $1.60 \times 10^{-1}$  5th, 2024

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(c) Three Mole (d) Four Moles (4) Which Of The Following Weighs The Most? (a) One G-atom Of Nitrogen (b) One Mole Of Water (c) One Mole Of Sodium (d) One Molecule Of H<sub>2</sub>SO<sub>4</sub> (5) 5.0 Litre Of 5th, 2024

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1. If  $(1 - \sin^2\beta) - \cos^2\beta$  (i) 0 (ii) 1 (iii)  $\tan^2\beta$  (iv)  $\cos^2\beta$  2. Evaluate  $\tan^2 300^\circ \sin 300^\circ + \cos 600^\circ \sin 2900^\circ \tan 2600^\circ \dots$  5th, 2024

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