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KEY Chemistry: PH And POH Calculations Part 1: Fill In The Missing Information In The Table Below.

	PH [ H 3 O 1+]	POH [ OH1-]	ACID Or BASE?
3.78	$1.66 \times 10^{-4}$	M	10.22
6.03	$\times 10^{-11}$	M	
Acid	3.41	$3.89 \times 10^{-4}$	M
	10.59	$2.57 \times 10^{-11}$	M
Acid	8.81	$1.55 \times 10^{-9}$	M
	5.19	$6.46 \times 10^{-6}$	M
Base	8.69	$2.04 \times 10^{-9}$	M
	5.31	$4.88 \times 10^{-6}$	M
Base	8.46	$3.47 \times 10^{-9}$	M
	5.54	$2.88 \times 10^{-6}$	M

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