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Electronegativity Difference And Bond Character As The Electronegativity Difference Grows From 0.00 To >3.00, The Sharing Of Electrons Becomes More Skewed. First One Atom Becomes Partially Negative, -, While The Other Becomes Partially Positive, +, And The Bond Is Polarized, A Polar Covalent Bond. Then As The Sharing Becomes More And More Unequal, The Electrons Reside Only On One Atom. Jan 11th, 2024 Solubility - Electronegativity And Intermolecular Forces Dipole, Hydrogen

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