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An Approved Wiring Method. See Figure 1 For Connection Diagrams. 9. Mar 1th,

2024.

2024Calculating Energy For Changes Of PhaseGive The Values For Each Step. Show Your Work. Write The Formula. Substitute The Values. Use Dimensional Analysis. Write Units. Use Sig. Figs. 1. Heat The Ice From – 15 °C To 0 °C ____ G °C 2. Melt The Ice Ice At 0 °C ____ 3. Heat The Water From 0 °C To 100 °C ____ 4. Jan 1th, 2024Internal Energy, Heat And Phase Changes.Because The System Is Thermally Isolated (closed Thermos), The Hot Coffee Is The Only Source Of Energy, So It Must Provide The Necessary 4272 J Of Heat. 4272 J For The Coffee, This Is A Decrease In Energy Q To Cool Coffee = M Coffee C Water(20 °C-T Initial) T Initial =20°C- Q To Cool Mar 2th, 2024.

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