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C) Baking Soda ( $\text{NaHCO}_3$ ) D) Ammonia E) Ammonium Dichromate Solution: A) Sodium Sulphate Chemical Formula Is  $\text{Na}_2\text{SO}_4$  Atoms- 2 Sodium, One Sulphur And 4 Oxygen Atoms. B) Quick Lime Chemical Formula Is  $\text{CaO}$ . Atoms- One Calcium Atom And 1 Oxygen Atom. May 5th, 2024

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I) The Deliquescent Salt P Turns Yellow, On Dissolving, In Water, And Gives Reddish Brown Precipitate With Sodium Hydroxide Solution. Ii) The White Crystalline Solid Q Is Soluble In Water; It Liberates A Pungent Smelling Gas When Heated With Sodium Hydroxide Solution. Mar 9th, 2024

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4. Conducts Electricity When It Is In The Molten State (iv) To Increase The PH Value Of A Neutral Solution, We Should Add: 1. An Acid 2. An Acid Salt 3. An Alkali 4. A Salt (v) Anhydrous Iron(III) Chloride Is Prepared By: 1. Direct Combination 2. Simple Displacement 3. ... Jan 6th, 2024

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The Deliquescent Salt P Turns Yellow On Dissolving In Water And Gives A Reddish

Brown Precipitate With Sodium Hydroxide Solution. li. The White Crystalline Solid Q Is Soluble In Water. It Liberates A Pungent Smelling Gas ... Lead Bromide Conducts Electricity. li. Copper Reacts With Nitric Acid To Produce Nitrogen Dioxide. May 3th, 2024

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(a) Definite Volume, No Definite Shape, Highly Compressible, Least Rigid (b) No Definite Volume, No Definite Shape, Highly Compressible, Least Rigid (c) Definite Volume, Definite Shape, Highly Compressible, Least Rigid (d) Definite Volume, No Definite Shape, Highly Compressible, Highly Rigid. 4 Feb 2th, 2024

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(ii) What Charge Will Y Have After The Addition Of Electros? (iii) Which Electrode Will Y Migrate To During The Process Of Electrolysis? 14. Answer The Following: (i) Acids Dissolve In Water And Produce Positively Charged Ion. Draw The Structure Of These Positive Ions. (ii) Exp Mar 7th, 2024

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Types (i) Physical Change (ii) Chemical Change. Physical Change A Physical Change Is A Temporary Change In Which No New Substance Is Formed And Chemical Composition Remains The Same. ... Original State After A Change, It Is Said To Be An Irreversible Change. Example: Souring Of Milk. Question 3. Menti Apr 4th, 2024

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(iii) If The Empirical Formula Of A Compound Is CH And It Has A Vapour Density Of 13, Find The Molecular Formula Of The Compound. [1] (d) State One Relevant Observation For Each Of The Following: (i) When Crystals Of Copper Nitrate Are Heated In A Test Tube. (ii) When The Gaseous P Jan 8th, 2024

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(c) Calculate The Empirical Formula Of A Compound Having The Following Percentage Composition. [Na = 42.1%, P = 18.9% And Oxygen = 39%] 23, P = 31 And O = 16]. [3] [4] In The Spaces : When The Sodium Atom Reacts With Chlorine Atom To Form Sodium Chloride, The Sodium Atom An Electron To Apr 10th, 2024

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