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(2) To Determine The Proper Number Of Sig. Figs When Multiplying Or Dividing The

Measurement With The Least Number Of Sig. Figs Is Boss G. If All This Is Confusing, Use The (A)tlantic – (P)acific Rule. If The Decimal Point Is (A)bsent – Start Counting Significant Figures From The Atlantic 14th, 2024

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2. Element J Has 27 Protons And 32 Neutrons. Element L Has 27 Protons And 33 Neutrons. ______3. Element X Has 17 Protons And 18 Neutrons. Element Y Has 18 Protons And 17 Neutrons. ______4. Element Q Has 56 Protons And 81 Neutrons. Element F Has 56 Protons And 82 Neutrons. ______5. Element T Has An Atomic Number 20 And Mass Number 40. Element Z ... 1th, 2024

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D. Uranium-235: 92 Protons, 143 Neutrons E. Carbon-12: 6 Protons, 6 Neutrons 3. Most Of An Atom's Mass Is Concentrated In The Nucleus. The Number Of Electrons And Protons Is The Same But Electrons Are So Light They Contribute Very Little Mass. The Mass Of The Proton Is 1,835 Times The Mass Of The Electron. Neutrons Have A Bit More Mass Than ... 3th, 2024

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