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On The Reaction $Al^{3+} + Mg \rightarrow Al + Mg^{2+}$ The Half Reactions Are $Al \rightarrow Al^{3+} + 3e^{-}$ $E^{\circ} = 1.66\text{ V}$ $Mg \rightarrow Mg^{2+} + 2e^{-}$ $E^{\circ} = 2.37\text{ V}$ Give The Balanced Cell Reaction And Calculate E° For The Cell. (See Work On Page 845) Solution: 0.71 V 2. Using 7th, 2024

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