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Since 1 Mol Of  $\text{Na}_2\text{SO}_4$  Contains 2 Mol Of  $\text{Na}^+$ , Amount  $\text{Na}^+ = 2 \times 0.17606 \text{ Mol} =$   
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1 Meth Methane 2 Eth Ethane 3 Prop Propane 4 But Butane 5 Pent Pentane 6 Hex Hexane 7 Hept

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