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The Minimum Radius Occurs When $n=1$ And Is Referred To As The Bohr Radius, R_1 , And Is Given By $R_1 = \frac{a_0}{Z}$ Where $a_0 = \frac{4\pi\epsilon_0\hbar^2}{m_e e^2}$. (15)

In Bohr's Model, The Total Energy E_n Of The Electron As It Orbits The Proton Is The Sum Of The Kinetic Energy K_n Due To Its Motion And The Potential Energy P_n It Has Because It Is Electrically Bound To The Proton So That $E_n = -\frac{13.6 \text{ eV}}{n^2}$ Jan 3th, 2024

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