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32-2 Radioactivity Unstable Nuclei Can Either Decay Into A Stable Nucleus Of Different  $N$  And  $Z$ , Or Can Return To The Ground State From An Excited State. Three Different Types Of Decay Particles: 1. Alpha Particles, Which Consist Of Two Neutrons And Two Protons, And Are Nuclei Of 2. Electrons 20th, 2024

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Simplest Form, The Form Factor Is The Fourier Transform Of The Charge Density. For Low Momentum Transfer,  $F(q) \approx Z(1 - \frac{1}{6} Q^2 R_{rms}^2 + \dots)$ . (N.B. Actually Scattering Theory And Form Factors Can Be More Complicated Than This, But For The Simplest Cases The Above Are Approximately True.) F( 10th, 2024

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A. A  $^{62}\text{Ni}$  Atom Has More Mass B. The Protons, Neutrons, And Electrons Have More Mass C. They Have Exactly The Same Mass Binding Energy V The Atomic Mass Of  $^{62}\text{Ni}$  Is 61.928 u. Ni Consists Of 28 Protons, 34 Neutrons, And 28 Electrons. If We Add Up The Mass Of 28 Proton 20th, 2024

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3 2. This Question Is About Nuclear Reactions. (a) (i) Distinguish Between Fission And Radioactive Decay. (4) A Nucleus Of Uranium-235 ( $^{235}\text{U}$ ) May Absorb A Neutron And Then Undergo Fission To 16th, 2024

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