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Falling In Love With Him. Mar 1th, 2024.

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The Phone Company To Replace Her Mobile. 4 I ~nd It Easy To Misunderstand What
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Answer On Question #58182 Chemistry Organic Chemistry ...2) Aniline And Acetanilide Stand In A One-to-one Molar Relationship, So 0.0322 Mol Of Aniline Produces 0.0322 Mol Of Acetanilide. The Amount Of Acetic Anhydride Does Not Play A Role, Since It Is In Excess. Then, $n(\text{Aniline}) = n(\text{Acetanilide}) = 0.0322$. The Molar Mass Of Acetanilide: $(\text{C}_8\text{H}_9\text{NO}) = 135.17 \text{ g/mol}$. $0.0322 \text{ mol} \times 135.17 \text{ g/mol} = 4.35 \text{ g}$. Jan 2th, 2024

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