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Worksheet: Mixed Problems—Mole/Mole Name And ... $2 \text{Cu} + \text{O}_2 \rightarrow 2 \text{CuO}$  A. If 101 Grams Of Copper Is Used, How Many Moles Of Copper (II) Oxide Will Be Formed? B. If 5.25 Moles Of Copper Are Used, How Many Moles Of Oxygen Must Also Be Used? C. If 78.2 Grams Of Oxygen React With Copper, How Many Moles Of Copper (II) Oxide Will Be Produced? 2.  $\text{C}_4\text{H}_{10} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$  A. How Many Moles Of Butane ... Mar 1th, 2024

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