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Colligative Properties For Molecular Solute Vs. Ionic Solute. Example: 1.0 M Of Sugar & 1.0 M Of Salt Show Different ΔT B And ΔT F Even Though The Molality Concentrations Of The Solute Are Identical. Note That; ΔT = M • K F,b Where K Is Only A Function Of The Solvent !!! Reason For The Difference In Colligative Properties Must Be Based On ... Mar 1th, 2024

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