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Oxidation Number Of Each Of The Elements That Is Underlined. A)  $\text{NH}_3$   $\underline{-3}$  B)  $\text{H}_2\text{SO}_4$   $\underline{6}$  C)  $\text{ZnCO}_3$   $\underline{4}$  D)  $\text{Al}(\text{OH})_3$   $\underline{3}$  E)  $\text{Na}$   $\underline{0}$  F)  $\text{Cl}_2$   $\underline{0}$  17. Balance The Redox Equation Using The Half Reaction Method.  $\text{Al} + 3\text{Ag}^+ \rightarrow \text{Al}^{3+} + 3\text{Ag}$  18. Ci 1th, 2024

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