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Aug 25, 2014 · C) A Mixture May Be Either Homogeneous Or Heterogeneous. D) All Of The Above Statements Are True. 50) How Would You Classify Salt Water? 50) A) Mixture - Homogeneous B) Mixture - Heterogeneous C) Pure Substance - Compound D) Pure Substance - Element 51) How Would You Classify Raisin Bran? 51) A) Pure Substance - Element B) 2th, 2022

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Chemical Changes Do Not Affect The Amount Of Matter Present. Chemical And Physical Changes Are Always Accompanied By Energy Changes. Energy May Be Released Or Absorbed. Exothermic - A Process That Releases Heat Endothermic - A Process That Absorbs Heat 3th, 2022

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15.Which Particle Diagram Represents A Mixture Of An Element And A Compound?

A)are Heterogeneous B)are Homogeneous C)combine In A Definite Ratio D)consist Of Two Or More Substances 16.One Similarity Between All Mixtures And Compounds Is That Both A)The Mixture Is Homogeneous And Can Be Separated By Filtration. B)The Mixture Is Homogeneous And ... 2th, 2022

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Elements In The Same Period Have The Same Number Of Electron Energy Levels Or Orbitals. The Vertical ... An Atom Is The Smallest Neutral Particle Of An Element That Can Exist And Still Have All The Properties Of That Element. Atoms, In Turn, Consist Of Smaller Particles: Protons, Neutrons A 1th, 2022

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