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To Thesum Ofenergyin Form Heat Exchanged Between System And Surroundings (q) And The Work Done On The System (w). Δ Esystem = Q + W But The Laws Of Thermodynamics Are Far More Profound Than Simply Describing The Efficiency Of Steam Engines. T May 1th, 2024

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Section 17.2 Measuring And Expressing Enthalpy Changes (pages 511-517) This Section Explains How To Construct Equations And Perform Calculations That Show Enthalpy Changes For Chemical And Physical Processes. ~ Calorimetry (pages Jan 15th, 2024

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