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4. $\text{HCl(aq)} + \text{Sr(OH)}_2\text{(aq)} \rightarrow \text{SrCl}_2\text{(aq)} + \text{H}_2\text{O(l)}$
5. $\text{AlBr}_3\text{(s)} \rightarrow \text{Al(s)} + \text{Br}_2\text{(l)}$
Classification Of Chemical Reactions
Reactions, Much Like Elements And Compounds, Can Be Classified According To Type. The Ability To Recognize And Classify Reactions Can Help Us Predict
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 $\text{O}_2 \rightarrow 2\text{BaO}$
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Carbon Dioxide Gas (CO_2) Is Present, It Will Put The Flame Out. Note Any Change In The Appearance Of The Residue In The Test Tube. DISPOSAL: This May Safely Go Down The D 7th, 2024LAB: TYPES OF CHEMICAL REACTIONSE. Turn On The Methane Gas And Light The Bunsen Burner. Open The Air Holes To Add More Oxygen And Make A Blue Cone. (Use This Flame For The Next Trial.) Observation Word Equation Balanced Formula Equation Type Of Reaction F. Place A Small 7th, 2024Name: Types Of Chemical Reactions Lab IntroductionTypes Of Chemical Reactions Lab Introduction In This Unit, You Have Studied Various Types Of Chemical Reactions. Now You Will Transfer This Theoretical Knowledge Of Chemical Reactions To The Laboratory. In This Activity You Will Observe And Predict Products For Four Types Of Chemical Reactions 14th, 2024.

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