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Prof. Peter S. Shawhan, Room 4205B In The Physics Building, Pshawhan@umd.edu, 301-405-1580 If You Have A Question Or Issue That Can't Be Handled During Office Hours, Please Email Or Call Me. Teaching Assistants: Anirban Gangopadhyay, Guil Miranda, Conner Ro Jan 6th, 2024

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Answers To Questions On Page 210: 30. SeO 3 Has VSEPR Notation AX 3E 0 So While It Contains Polar Bonds, It Is Trigonal Planar In Shape, Symmetrical And Non-polar. 31. N 2H 4 Has Non-polar Covalent Bonds Between The Two Nitrogen Atoms A Jan 2th, 2024

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